

Fluor-Free Superhydrophobic and Eco-Friendly Superoleophobic Textiles

 **UCLouvain**
Institute of Condensed Matter
and Nanosciences

Ronggang Cai, Karine Glinel,
Bernard Nysten, Alain M. Jonas



David De Smet, Myriam Vanneste

Interreg
France-Wallonie-Vlaanderen



GoToS3
DURATEX

Avec le soutien du Fonds européen
de Développement Régional
Met de steun van het Europees Fonds
voor Regionale Ontwikkeling



met de steun van
west-vlaanderen
de gedreven provincie

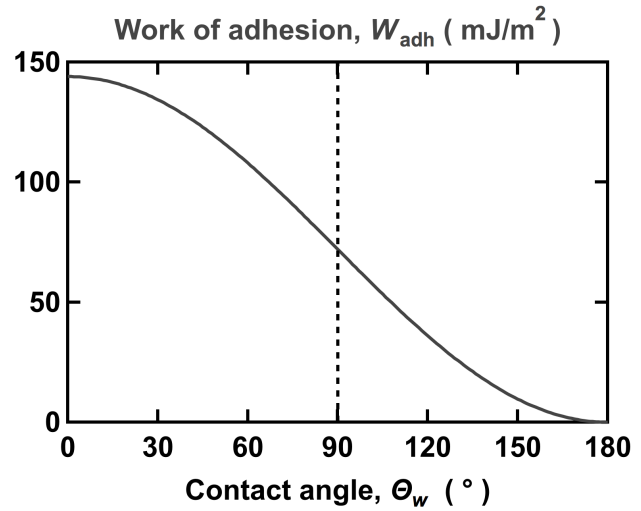
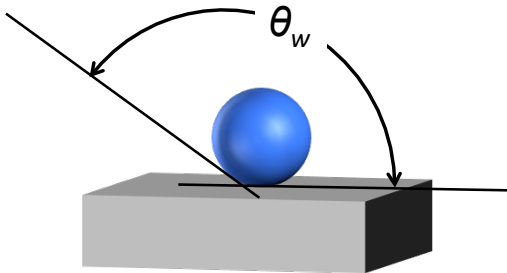


Why were long chain
perfluoroalkyls used
for water repellence?

The contact angle of a water droplet provides its work of adhesion on a flat surface, W_{adh}

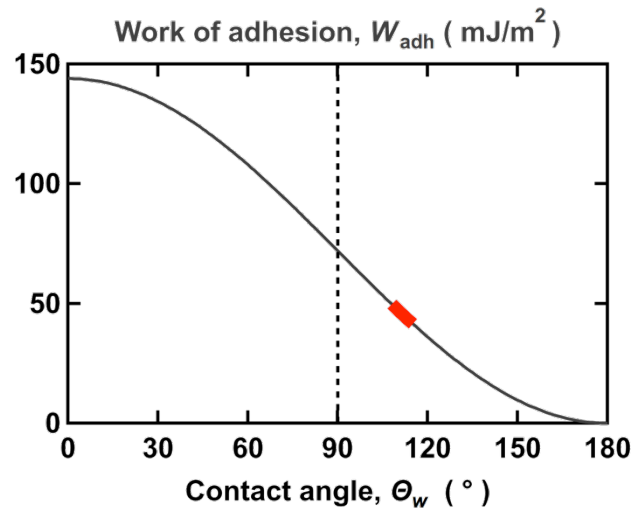
$$W_{adh} = \gamma_w (1 + \cos \theta_w)$$

$\sim 72 \text{ mJ/m}^2$



Long perfluoroalkyl chains result in a low work of adhesion and provide very good water repellence

Long perfluoroalkyl chains

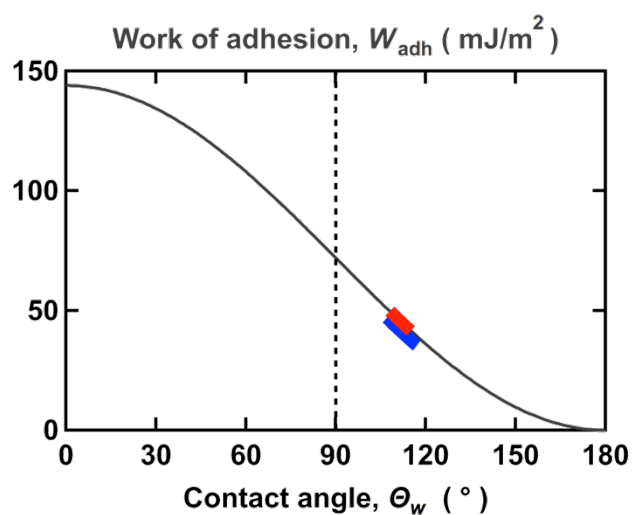


Are there possible
replacements for long chain
fluoroalkyls?

Other candidates are possible

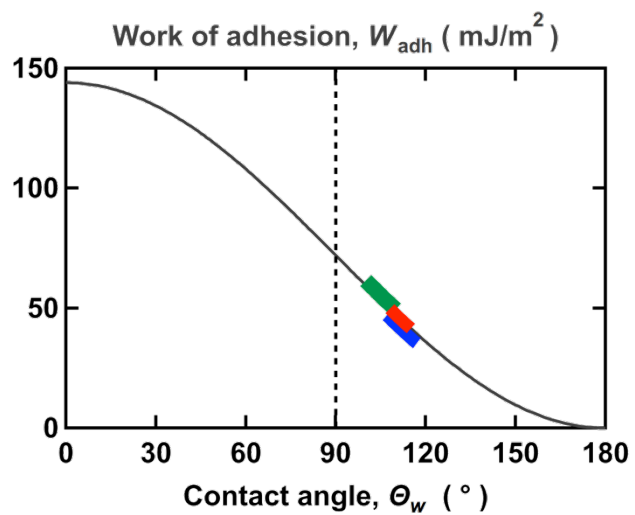
Silicones

Long perfluoroalkyl chains



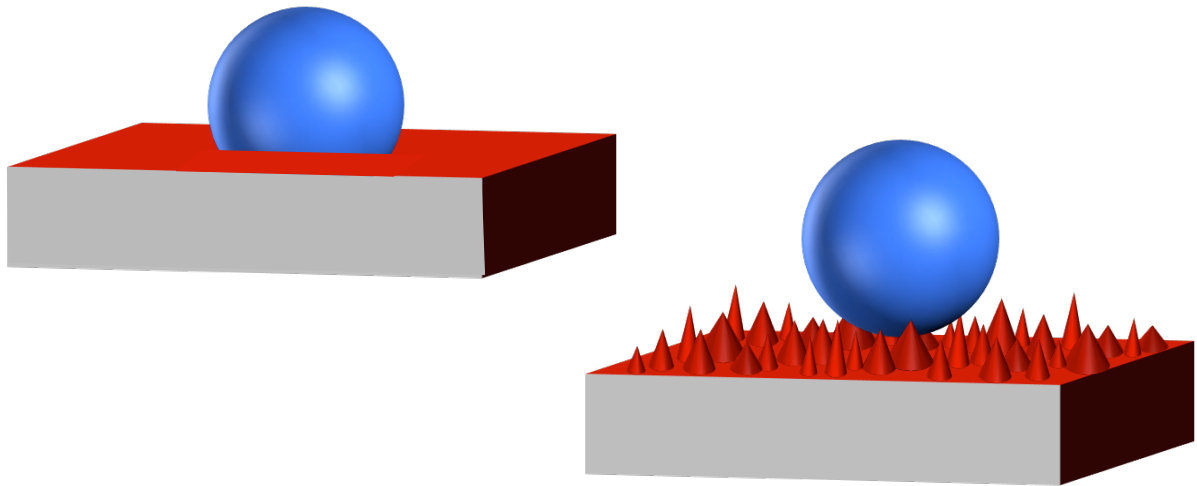
Other candidates are possible

Alkyl chains (waxes)
Silicones
Long perfluoroalkyl chains

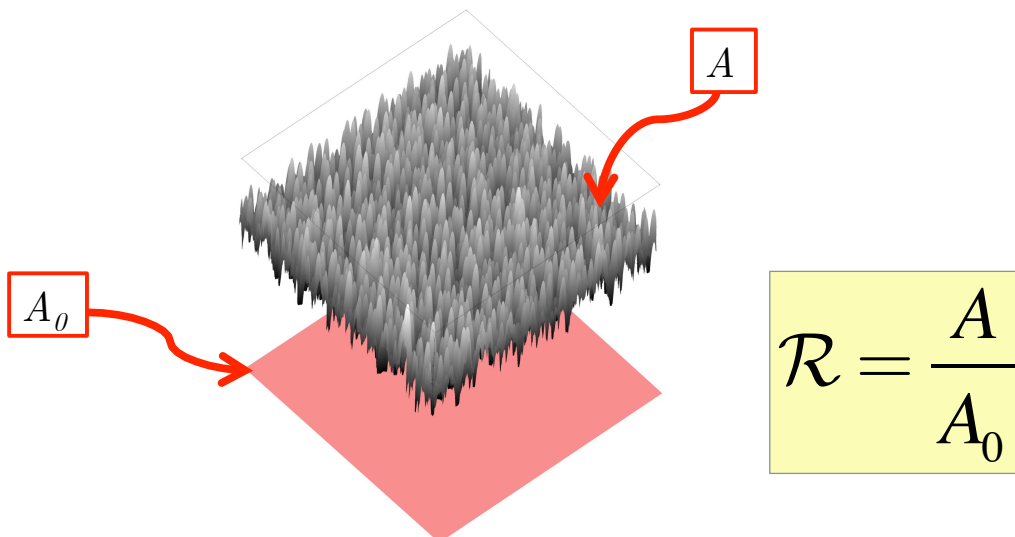


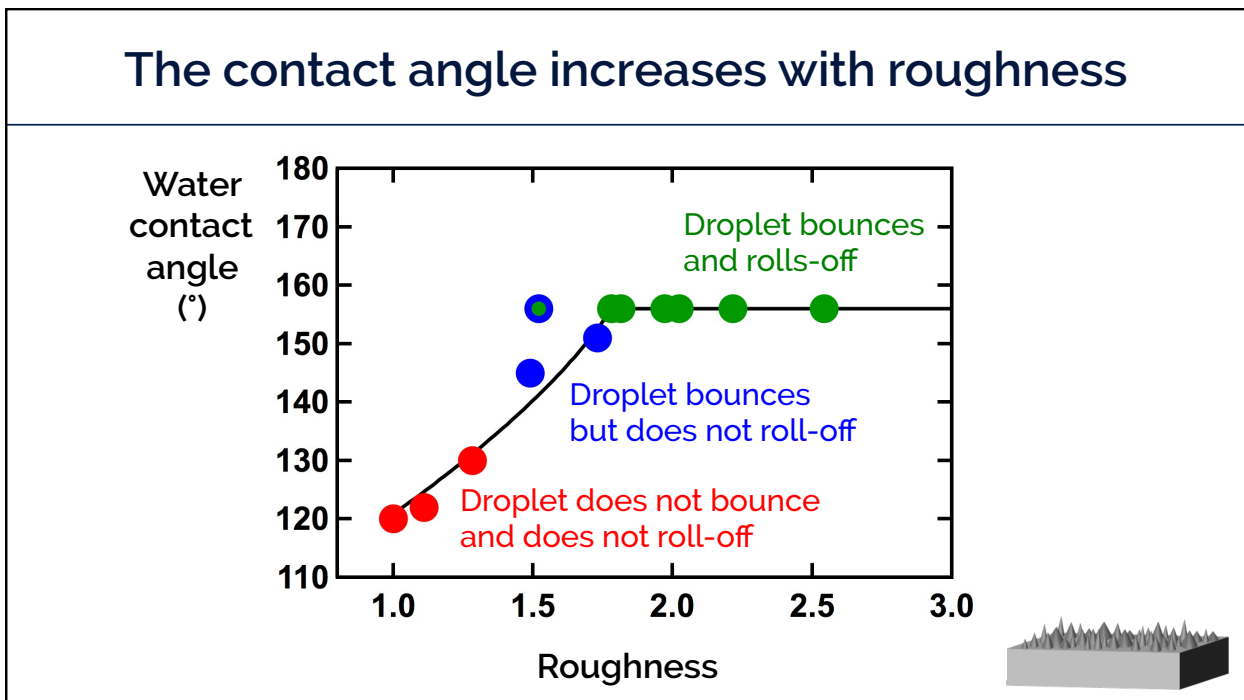
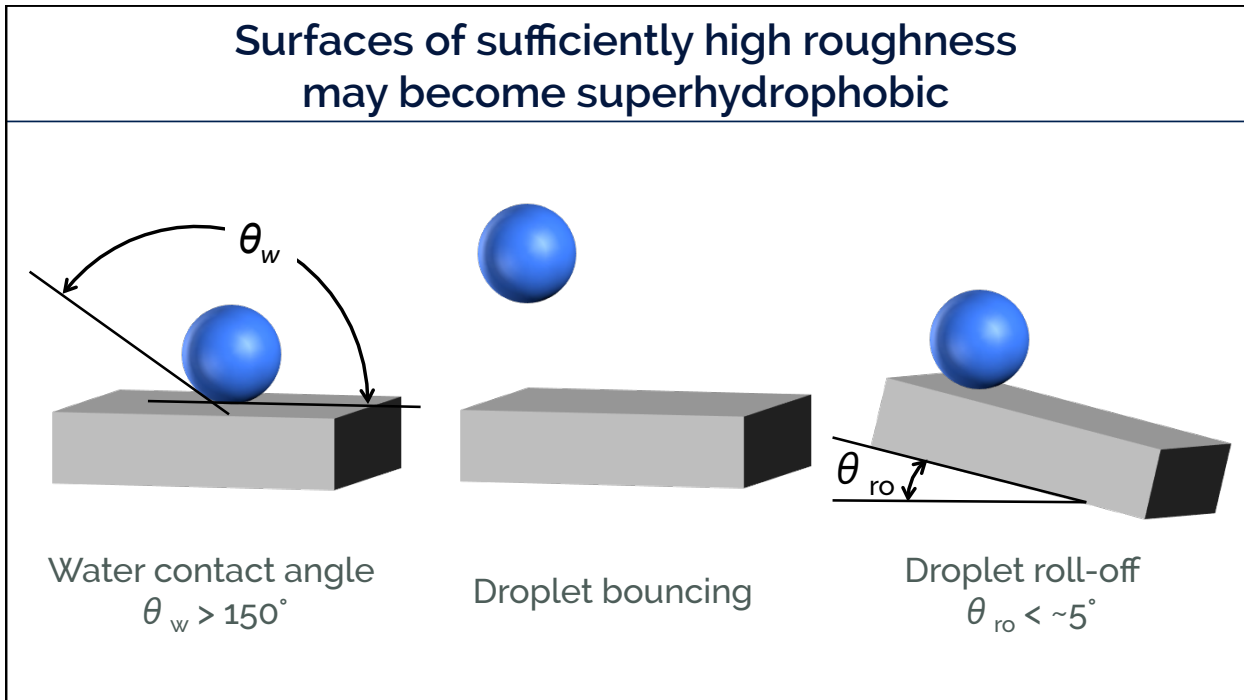
Chemistry is not the only
parameter one can play with

Surface roughness is another parameter controlling the contact angle

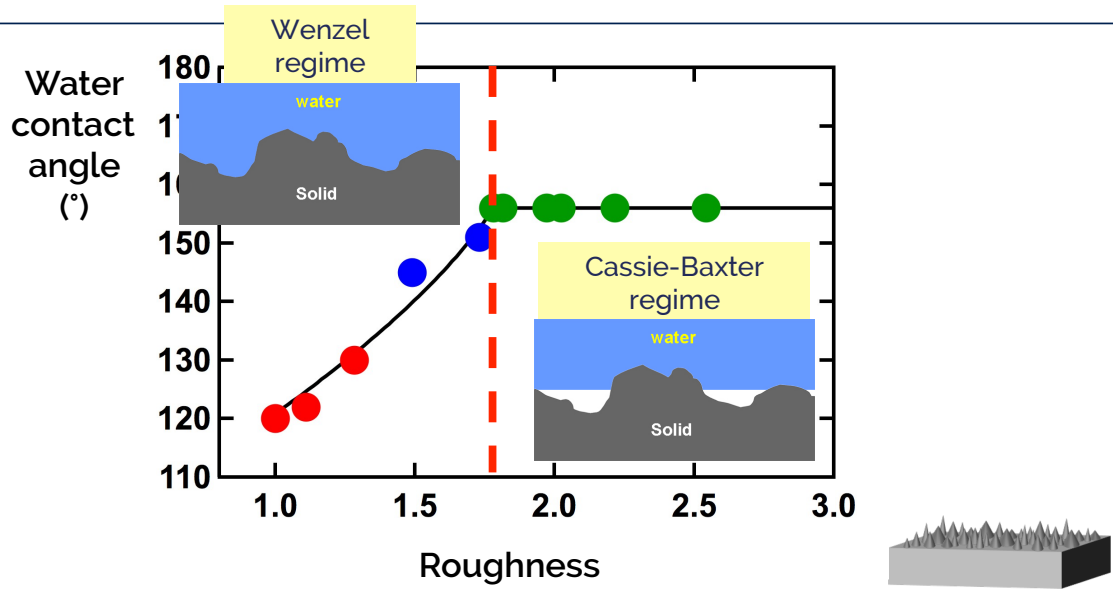


The roughness is defined as the increase of surface area



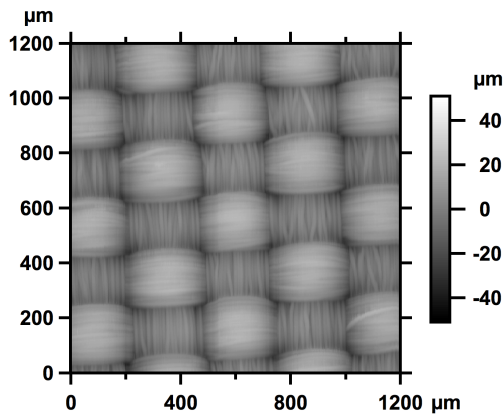


There exists a critical roughness for superhydrophobicity

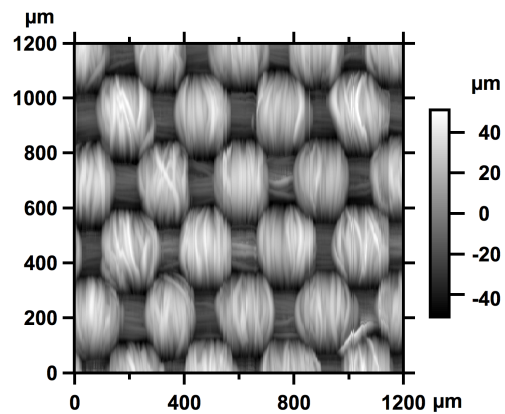


What is the role of fabric roughness in water repellence?

Woven fabrics have an intrinsic roughness which can be measured by profilometry

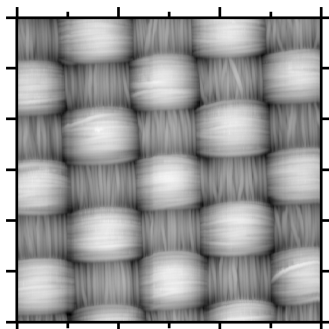


$$\mathcal{R} = 1.15$$



$$\mathcal{R} = 1.57$$

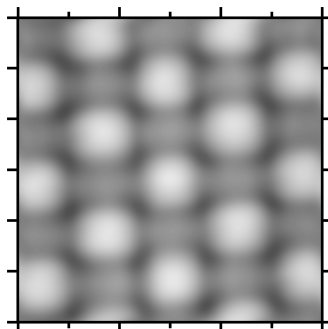
A fabric roughness arises from the weave pattern and the fiber packing in the yarns



Fabric pattern

$$\mathcal{R}$$

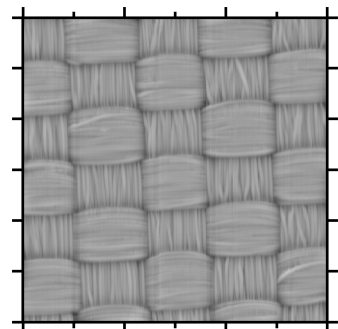
=



Weave pattern

$$\mathcal{R}_W$$

+

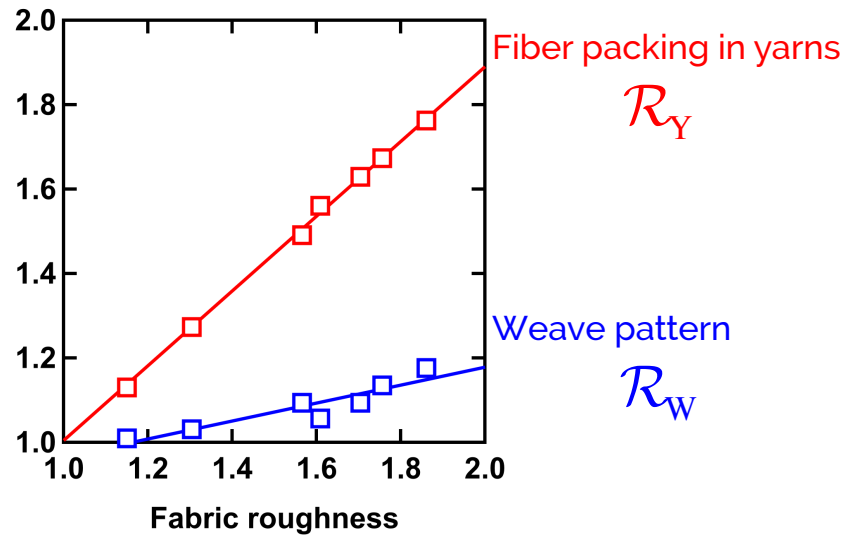


Fiber packing in yarns

$$\mathcal{R}_Y$$

$$\mathcal{R} = \mathcal{R}_W + \mathcal{R}_Y - 1$$

The fiber-in-yarn roughness dominates the roughness of woven fabrics



Is the intrinsic fabric roughness high enough to provide superhydrophobicity?

10 fabrics selected

roughness

1.85



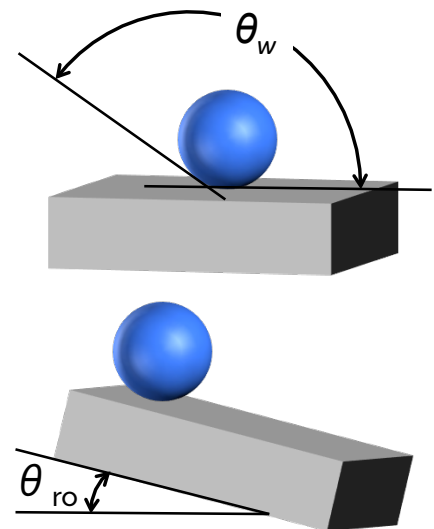
1.15

Dip-coated by 5 aqueous formulations

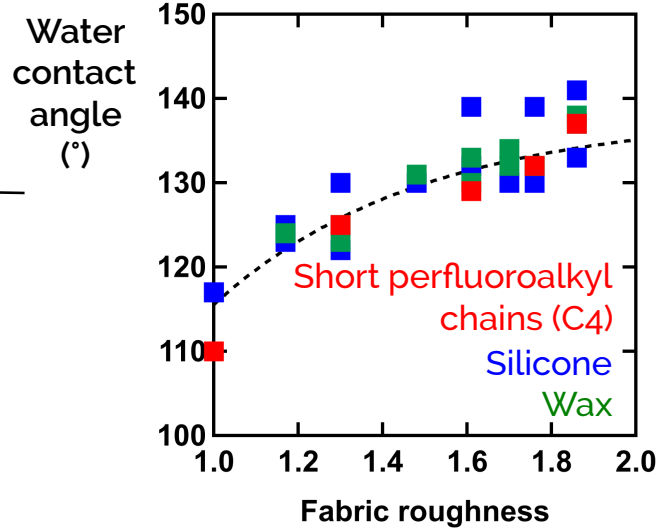
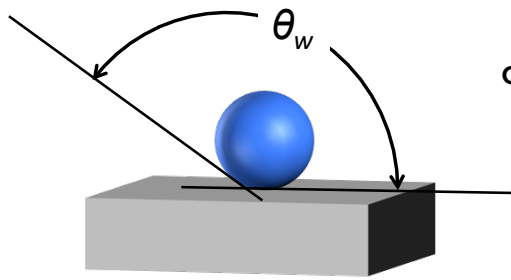
Short perfluoroalkyl chains (C4)

Silicone

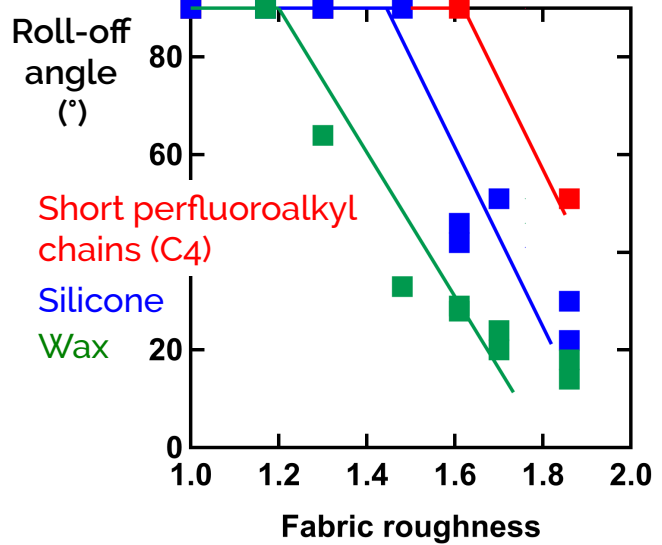
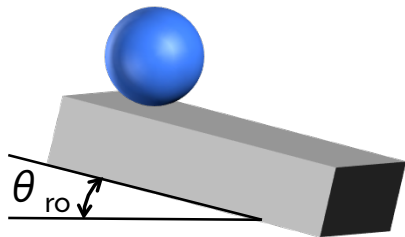
Wax



The contact angle increases with fabric roughness but does not discriminate between different coatings

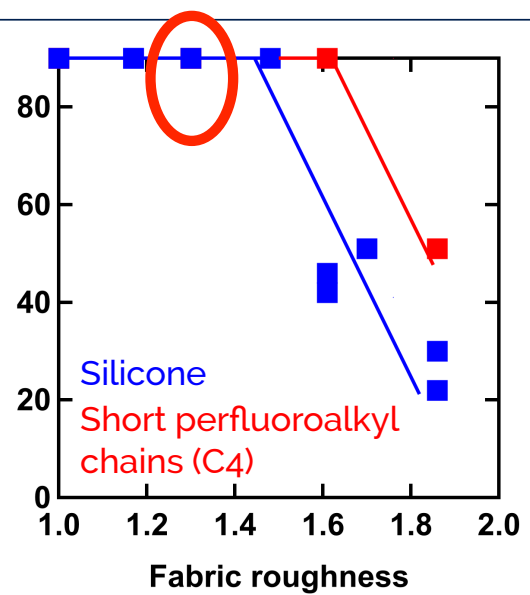
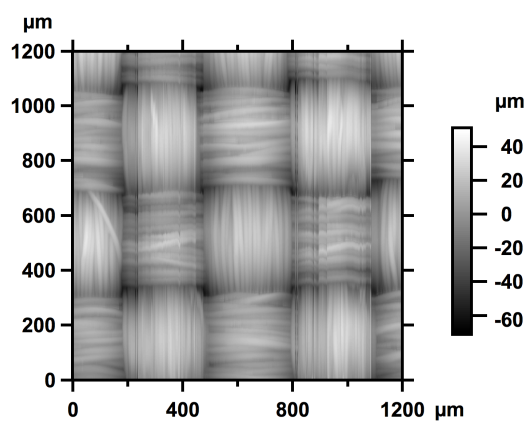


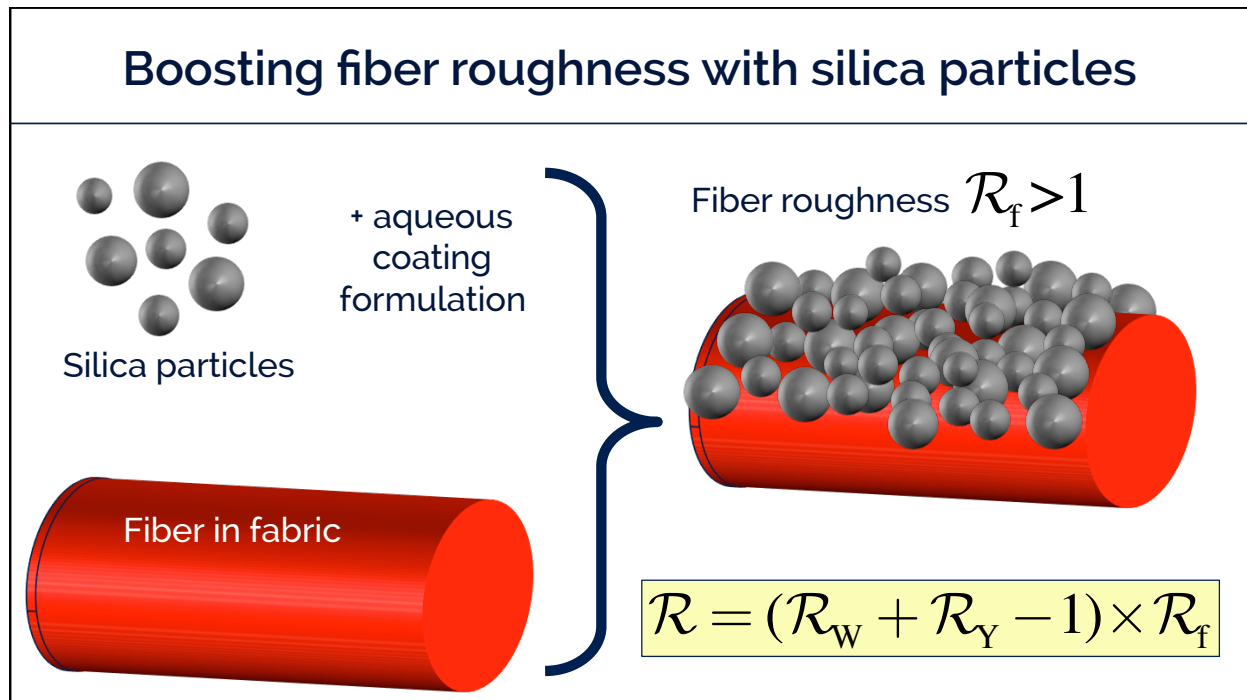
The roll-off angle decreases with fabric roughness and discriminates between different coatings



Can we improve the hydrophobicity of a fabric of low roughness?

We selected a fabric of low roughness (1.3)



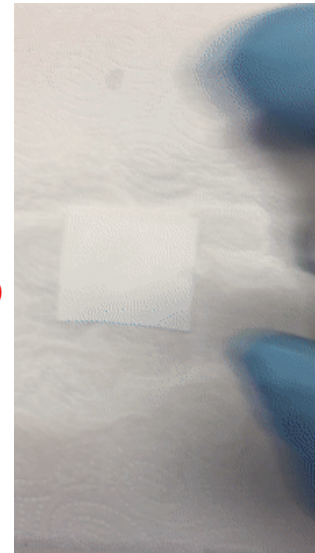
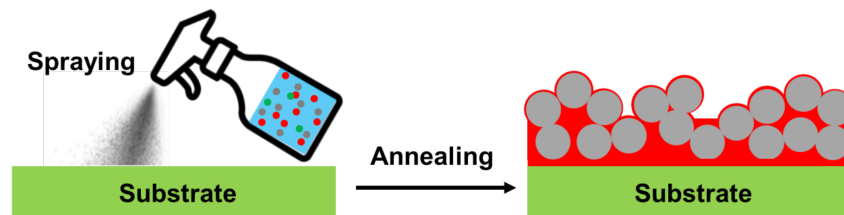


Results with a polyurethane having
short perfluoroalkyl chains

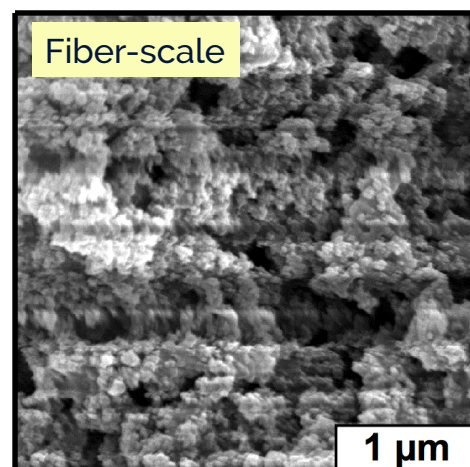
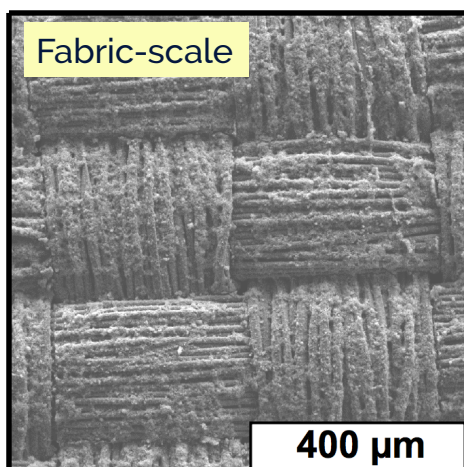
(aqueous emulsion, PM900, 3M)

Spraying the one-pot aqueous formulation

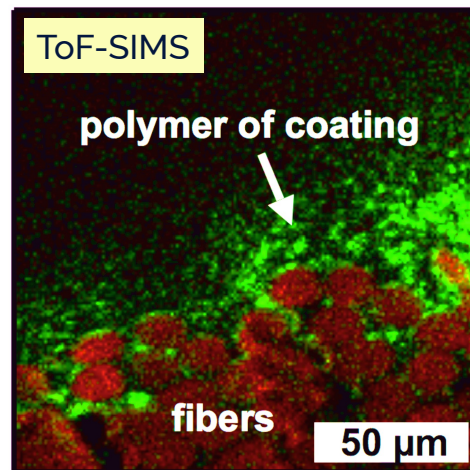
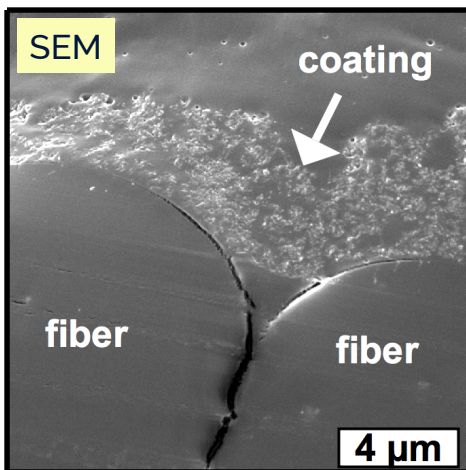
- Silica particles
- C4 fluorinated polymer
- Extender
- Water



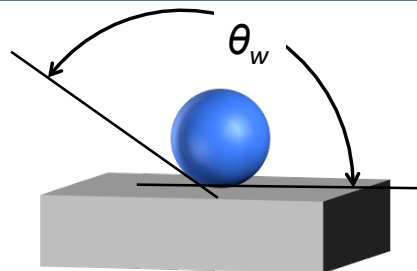
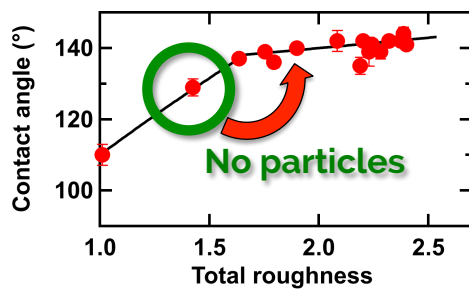
A typical resulting microstructure seen by SEM



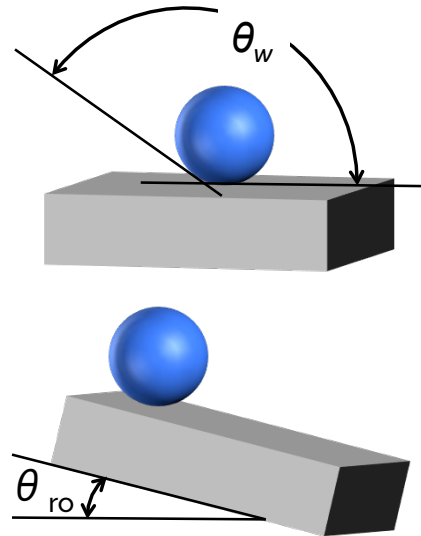
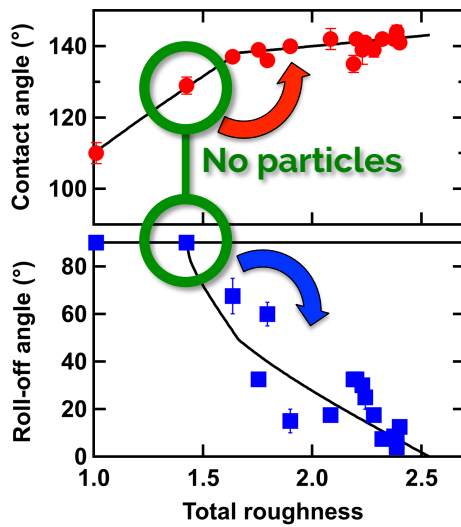
A typical transverse cut



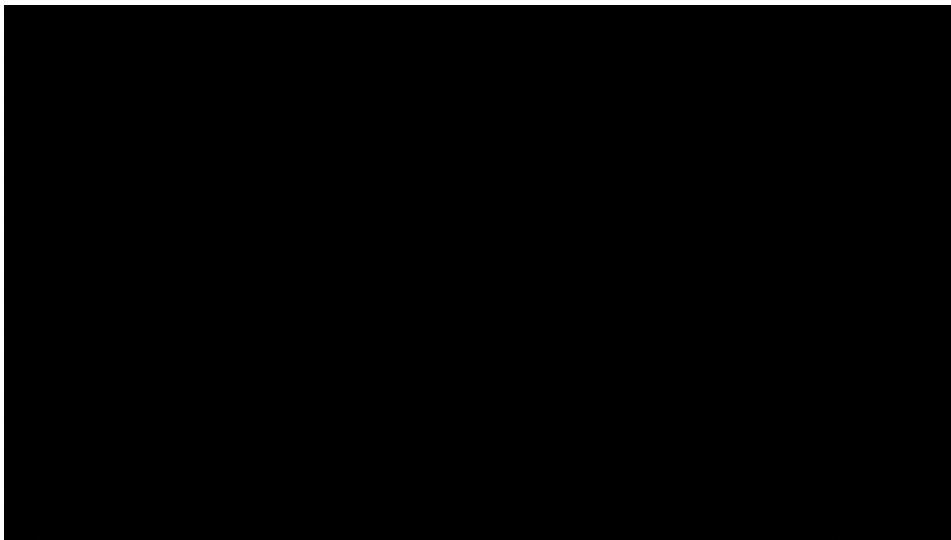
Increasing the roughness of the fibers results in increased water repellence



Increasing the roughness of the fibers results in increased water repellence



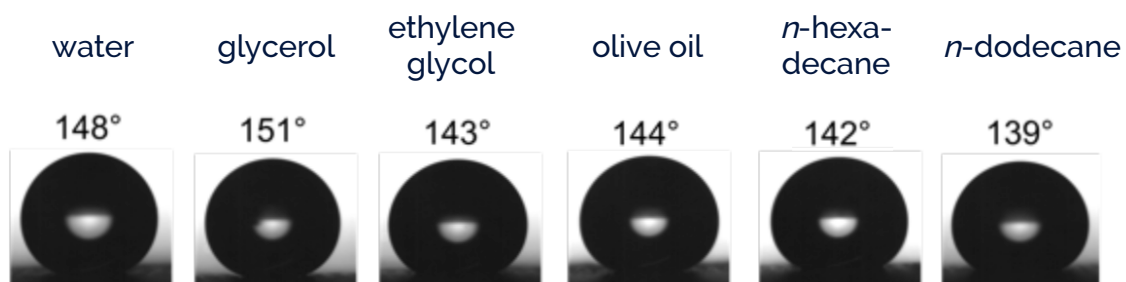
A superhydrophobic C4-based sample in action



The short C₄ perfluoroalkyl chains
also lead to superoleophobicity



The short C₄ perfluoroalkyl chains
also provide resistance to some organic solvents



However, shorter perfluoroalkyls also raise concerns

THE NEXT GENERATION
Industry shifted to shorter-chain PFAs and more complex structures; less is known about the safety risks of these molecules.

PFBS
Variations in chain length and branching produce dozens of variant structures.

PFHxS

FLUORINE Environmental Sciences Europe

Brendel et al. *Environ Sci Eur* (2018) 30:9
<https://doi.org/10.1186/s12302-018-0134-4>

RESEARCH

Short-chain perfluoroalkyl acids: environmental concerns and a regulatory strategy under REACH

Stephan Brendel, Éva Fetter, Claudia Staude, Lena Vierke and Annegret Biegel-Engler

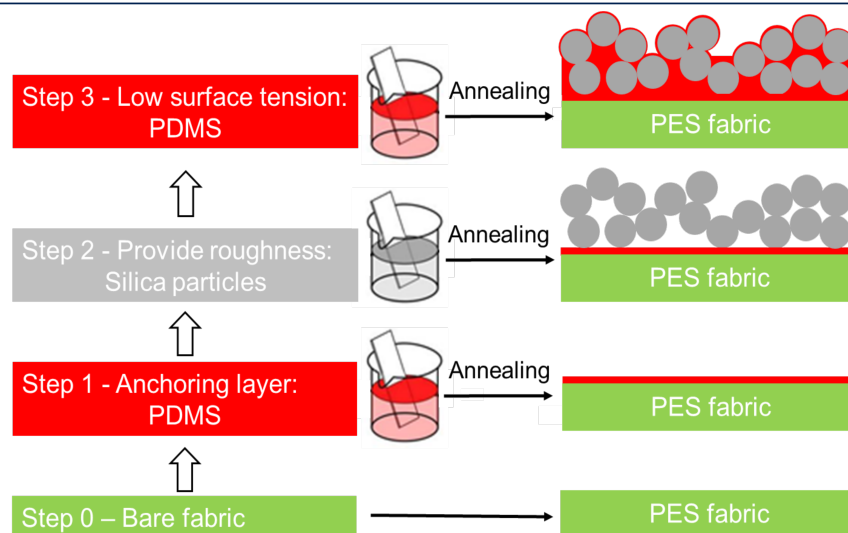
17 FEBRUARY 2019

MYSTERY COMPOUND
Researchers think the environment — with various PFAs

Results obtained with
a crosslinked silicone rubber

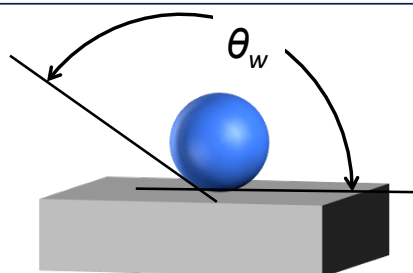
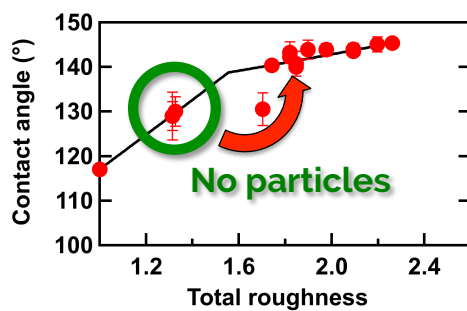
(aqueous emulsion, HC303, Wacker)

Silicone-based superhydrophobic coatings (dip-coating from aqueous suspensions)

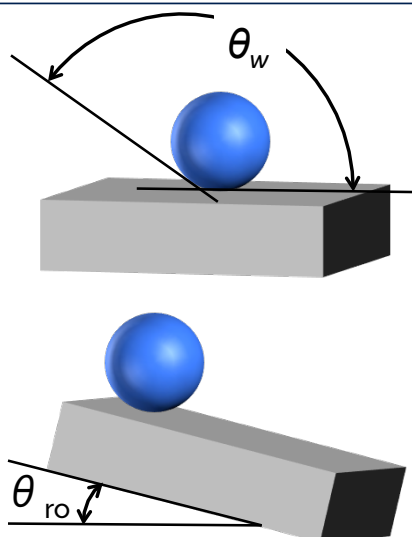
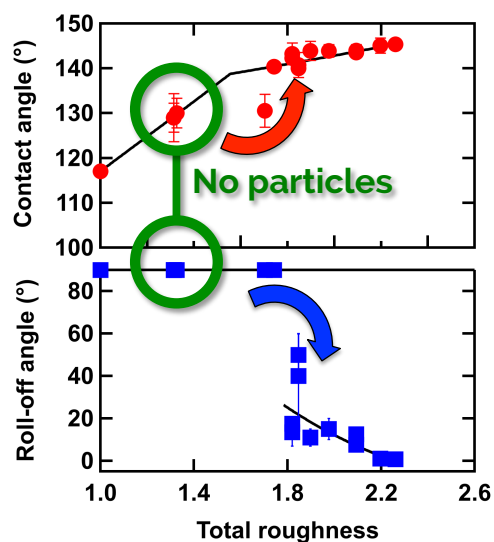


ACS Applied Materials & Interfaces (2018, 10, 15346–15351)

Again, increasing the roughness of the fibers results in increased water repellence



Again, increasing the roughness of the fibers results in increased water repellence



Water repellence of a silicone-based superhydrophobic fabric prepared from aqueous suspensions




Silicones might also rise concerns in the public

ECHA: Siloxanes D4, D5, and D6 Classified as SVHC
Posted on 5 July 2018 by kftchemieservice

The Member State Committee (MSC) of the ECHA has approved the classification of D4, D5, and D6 as substances of very high concern (SVHC). The German Environmental Ministry submitted the required reports ([Annex XV Report on D4](#) and [Annex XV - Identification of D4 and D5 as SVHC in March of this year](#)). The ECHA published the [Report on D6](#).

The experts in the MSC rated D4 as persistent, bioaccumulative (it accumulates in the food chain), and toxic (PBT), but they assigned only persistent and bioaccumulative to D5 and D6. Nevertheless, D5 and D6 can also be classified as PBT when both show D4 impurities in a concentration equal to or greater than 0.1% by weight.

Silicone compounds D4 (cyclotetrasiloxane), D5 (cyclopentasiloxane), and D6 (cyclohexasiloxane) are often found in personal care products and flow into the environment along with waste water. They are also important source materials for certain silicones often found as residues in finished products.



Substance Name:
Octamethylcyclotetrasiloxane (D4)

EC Number: 209-136-7

CAS Number: 556-67-2

**MEMBER STATE COMMITTEE
SUPPORT DOCUMENT
FOR IDENTIFICATION OF
OCTAMETHYLCYCLOTETRAILOXANE (D4)
AS A SUBSTANCE OF VERY HIGH CONCERN
BECAUSE OF ITS PBT¹ AND vPvB² PROPERTIES
(ARTICLE 57D&E)**

Adopted on 13 June 2018

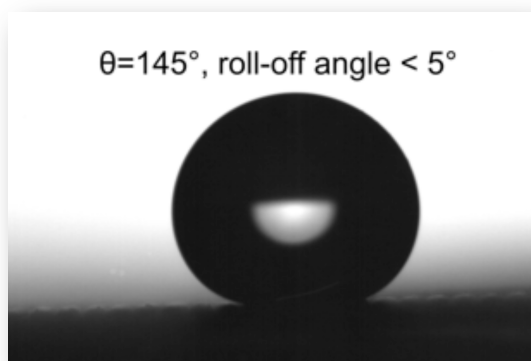
¹PBT means persistent, bioaccumulative and toxic.
²vPvB means very persistent and very bioaccumulative.

Wax-based coatings are used in nature



Preliminary studies indicate wax-based alternatives to be promising

Silicone replaced by water-based paraffin wax (Contraqua WE)
(total roughness: $1.3 \times 1.72 = 2.23$)



Main conclusions

1. Total roughness is a predictor of water repellence performance;
other parameters may have to be considered

2. Total roughness $\mathcal{R} = (\mathcal{R}_W + \mathcal{R}_Y - 1) \times \mathcal{R}_F$

\downarrow \downarrow \downarrow
 Weave pattern Fibers-in-yarn Fiber surface
 (limited) (important) (very important)

3. Fiber surface roughness can be boosted by silica nanoparticles;
other methods certainly exist

4. Different formulations can be discriminated by the roll-off angle
measured on a set of fabrics of different roughness

5. Our studies suggest: waxes > silicones > C4 perfluoroalkyls

Acknowledgments



GoToS3

DURATEX

Avec le soutien du Fonds européen
de Développement Régional
Met de steun van het Europees Fonds
voor Regionale Ontwikkeling

Special thanks to:

UCLouvain: Arnaud Delcorte,
Claude Poleunis, Cécile
D'Haese, Romain Vermeyen

Certech: Nicolas Mannu &
Benoît Kartheuser



Fluor-Free Superhydrophobic and Eco-Friendly Superoleophobic Textiles

UCLouvain

Institute of Condensed Matter
and Nanosciences

Ronggang Cai, Karine Glinel,
Bernard Nysten, Alain M. Jonas



David De Smet, Myriam Vanneste



GoToS3

DURATEX

Avec le soutien du Fonds européen
de Développement Régional
Met de steun van het Europees Fonds
voor Regionale Ontwikkeling

