

Playing with fabric roughness to improve water (and oil?) repellency



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DURATEX

Avec le soutien du Fonds européen
de Développement Régional
Met de steun van het Europees
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Wallonie

met de steun van

west-vlaanderen

de gedreven provincie



1

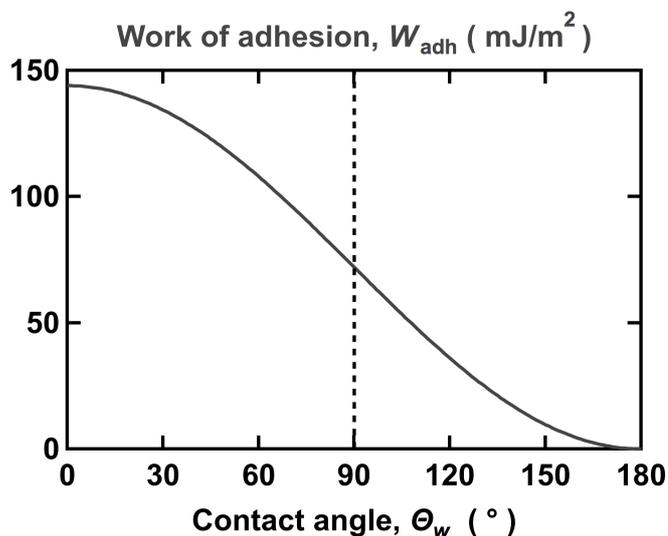
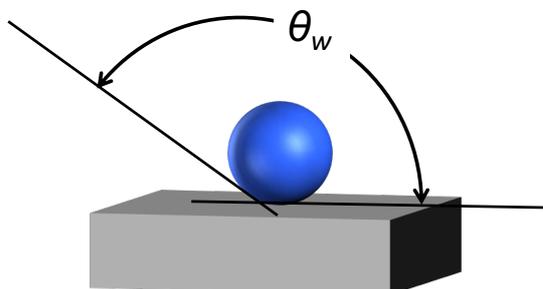
Why were long chain
perfluoroalkyls used
for water repellence?

2

The contact angle of a water droplet provides its work of adhesion on a flat surface, W_{adh}

$$W_{adh} = \gamma_w (1 + \cos \theta_w)$$

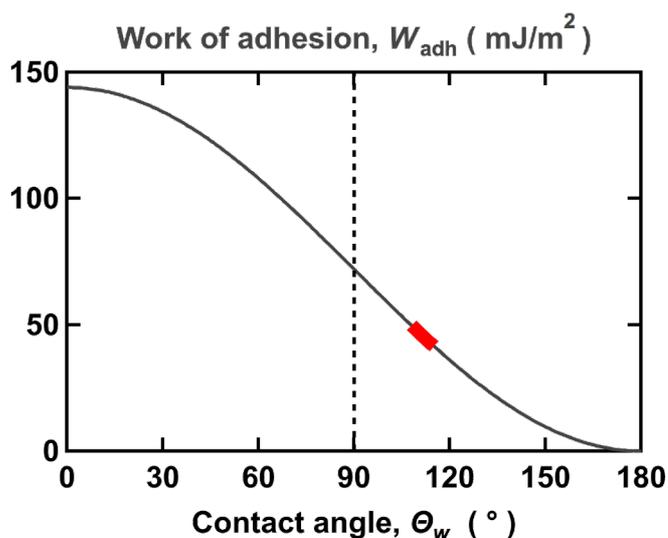
$\sim 72 \text{ mJ/m}^2$



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Long perfluoroalkyl chains result in a low work of adhesion and provide very good water repellence

Long perfluoroalkyl chains



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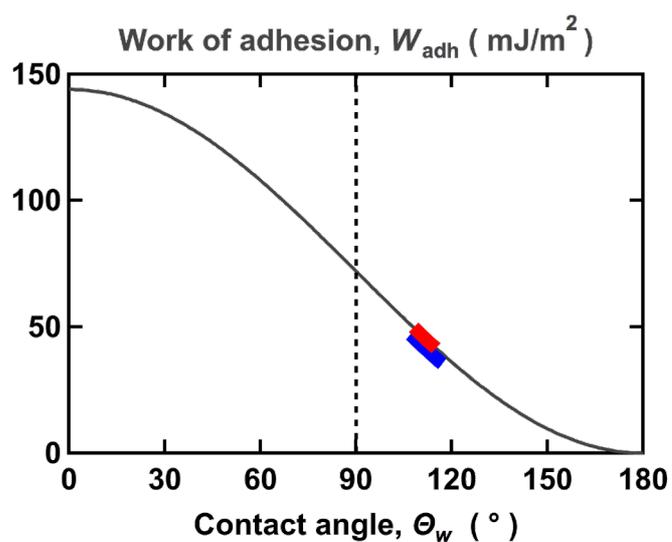
Are there possible replacements
for long chain fluoroalkyls?

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Other candidates are possible

Silicones

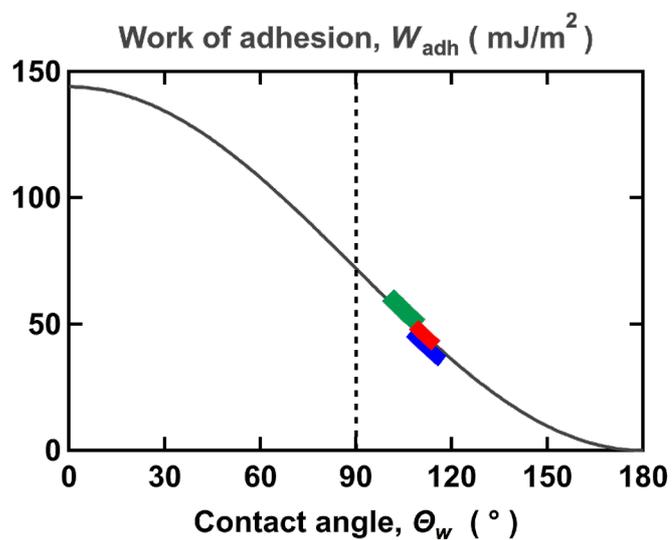
Long perfluoroalkyl chains



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Other candidates are possible

Alkyl chains (waxes)
Silicones
Long perfluoroalkyl chains

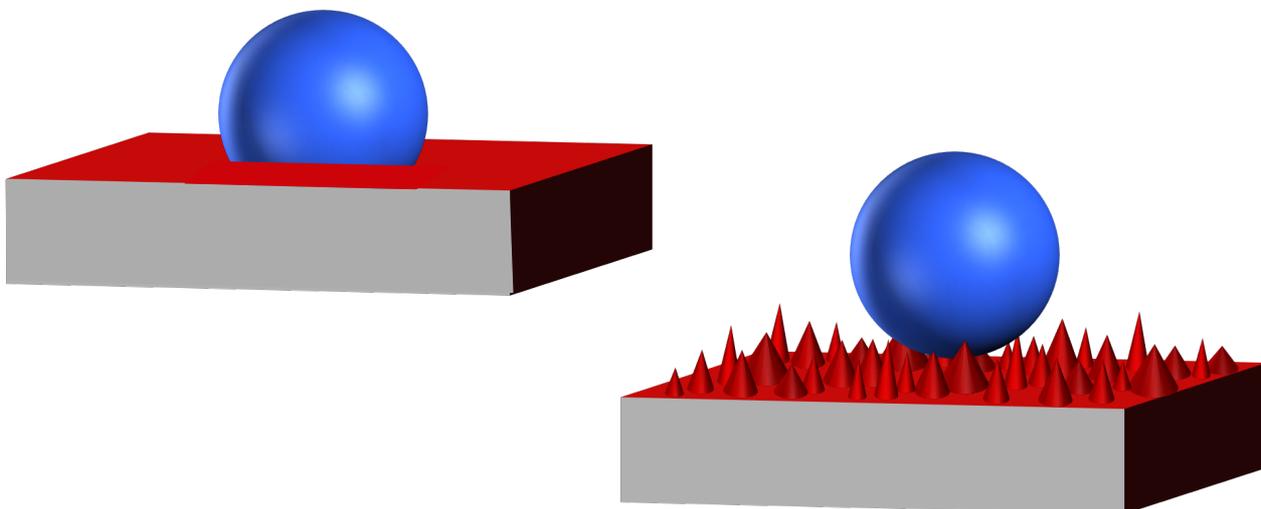


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Chemistry is not the only
parameter one can play with

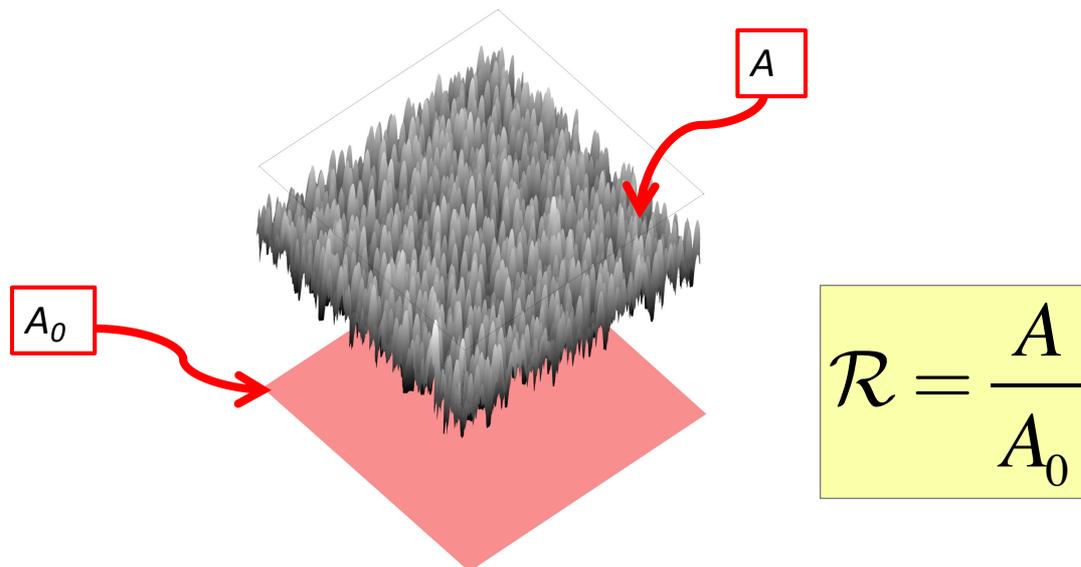
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Surface roughness is another parameter controlling the contact angle



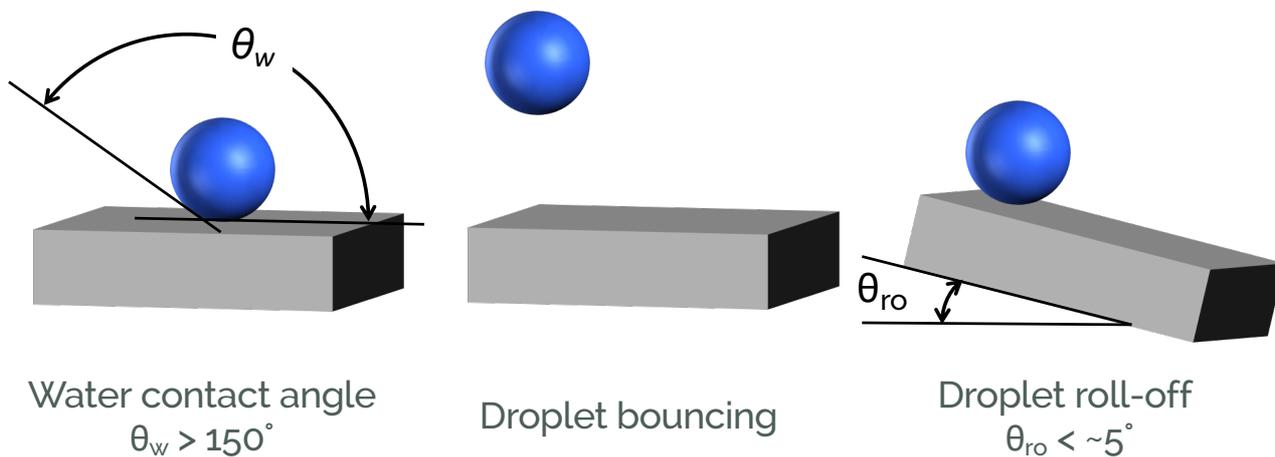
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The roughness is defined as the increase of surface area



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Surfaces of sufficiently high roughness
may become superhydrophobic

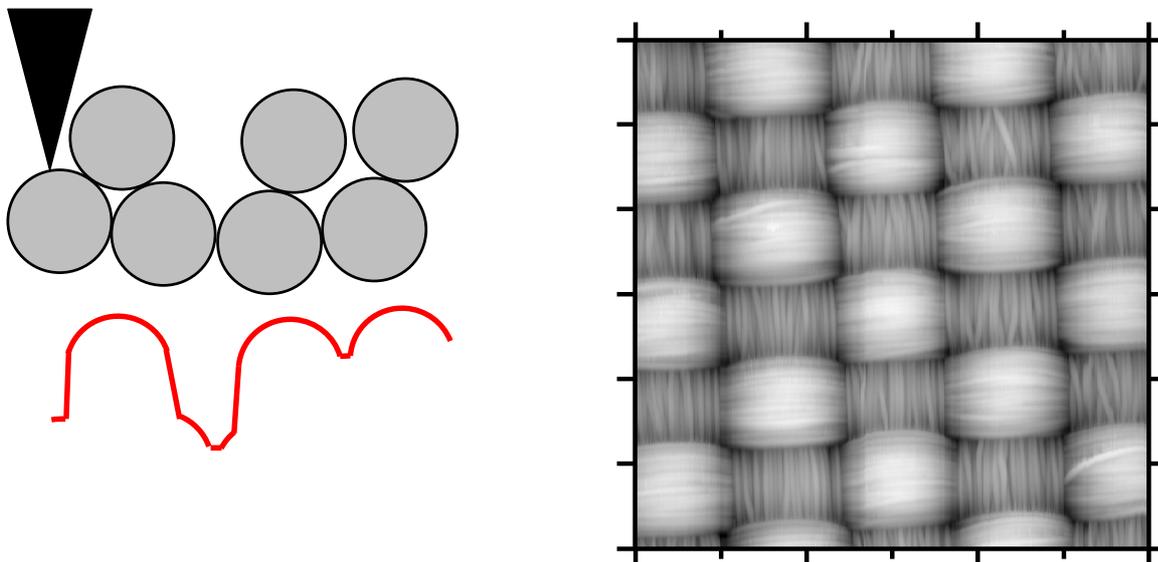


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What is the role of fabric
roughness in water repellence?

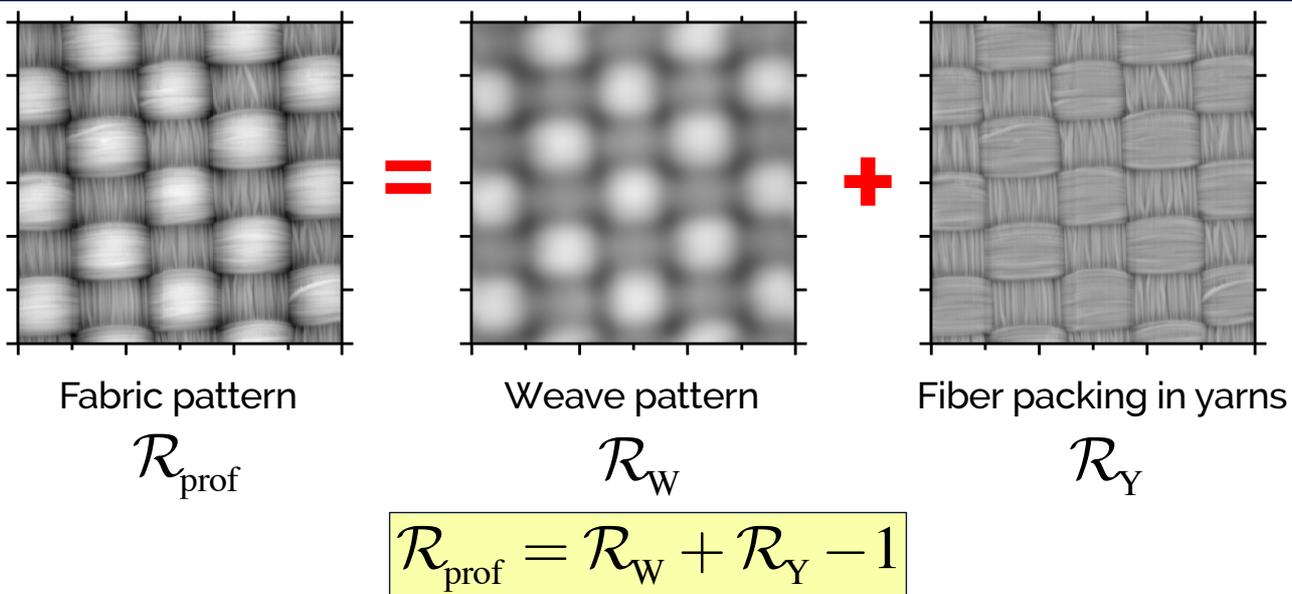
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An experimental roughness can be measured
by profilometry and AFM



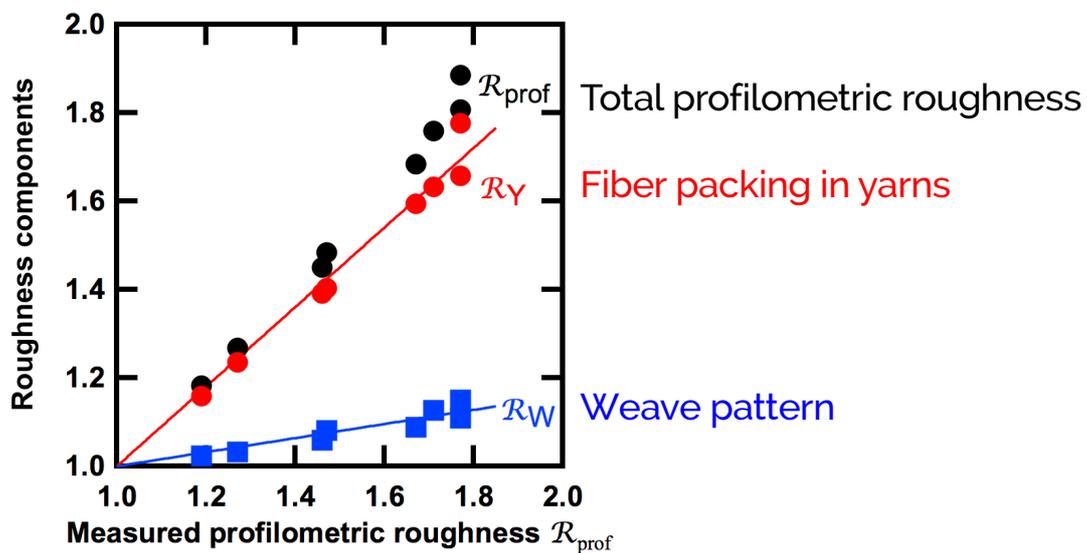
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The profilometry roughness arises from the weave pattern
and the fiber packing in the yarns



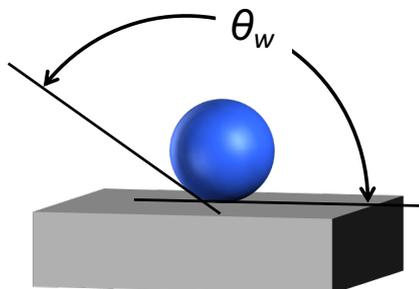
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The fiber-in-yarn roughness dominates the roughness of woven fabrics

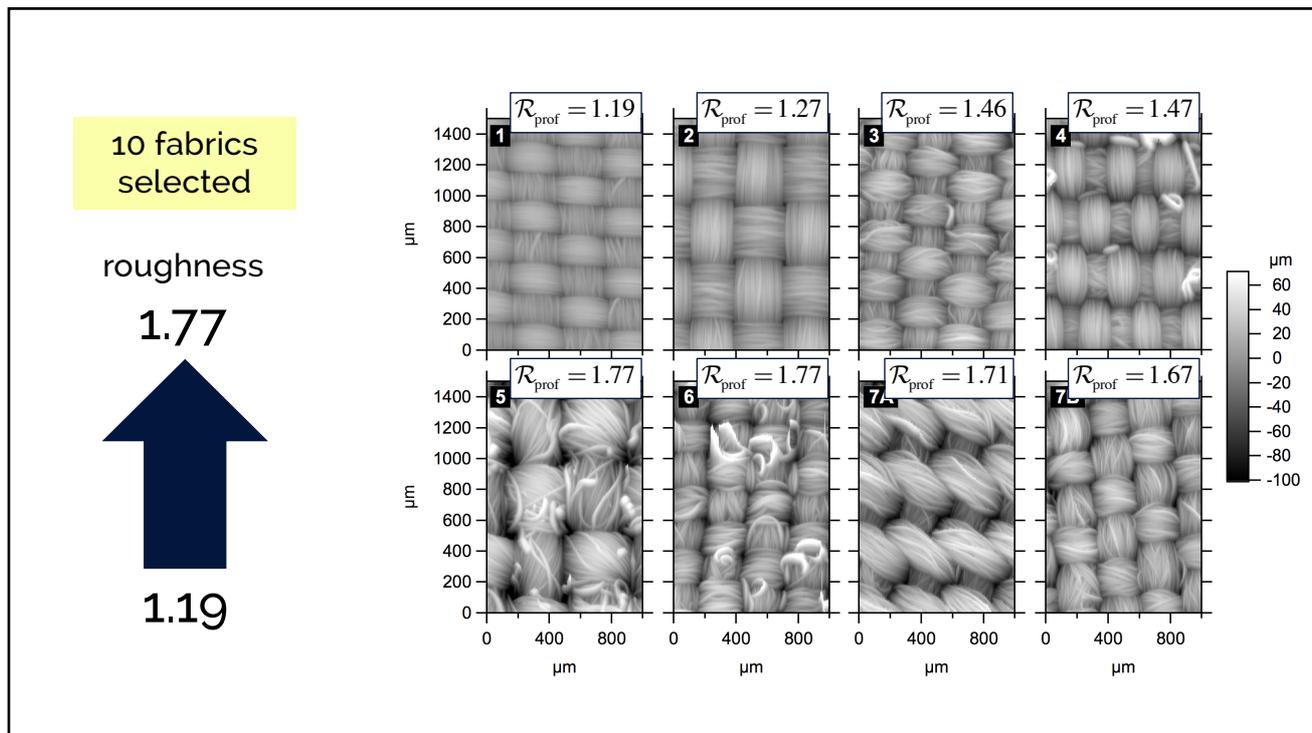


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Static contact angle of coated woven fabrics



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Three aqueous formulations of similar θ_w
are used to dip-coat our fabrics

	θ_w (H ₂ O, flat surface) (°)
Wax-modified melamin resin	110
Silicone rubber	111
Perfluorobutyl (C ₄) -modified PUR	110

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Short perfluoroalkyls raise concerns

Brendel *et al. Environ Sci Eur* (2018) 30:9
<https://doi.org/10.1186/s12302-018-0134-4>

 Environmental Sciences Europe

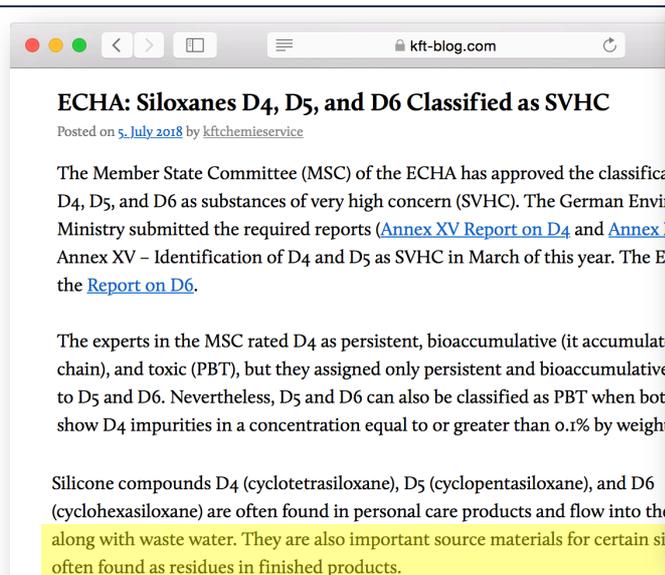
RESEARCH

Short-chain perfluoroalkyl acids: environmental concerns and a regulatory strategy under REACH

Stephan Brendel , Éva Fetter, Claudia Staude, Lena Vierke and Annegret Biegel-Engler

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Silicones might also rise concerns in the public



ECHA: Siloxanes D4, D5, and D6 Classified as SVHC
 Posted on 5 July 2018 by kftchemieservice

The Member State Committee (MSC) of the ECHA has approved the classification of D4, D5, and D6 as substances of very high concern (SVHC). The German Environmental Ministry submitted the required reports ([Annex XV Report on D4](#) and [Annex XV Report on D5](#)) as SVHC in March of this year. The ECHA also published the [Report on D6](#).

The experts in the MSC rated D4 as persistent, bioaccumulative (it accumulates in the food chain), and toxic (PBT), but they assigned only persistent and bioaccumulative to D5 and D6. Nevertheless, D5 and D6 can also be classified as PBT when both show D4 impurities in a concentration equal to or greater than 0.1% by weight.

Silicone compounds D4 (cyclotetrasiloxane), D5 (cyclopentasiloxane), and D6 (cyclohexasiloxane) are often found in personal care products and flow into the environment along with waste water. They are also important source materials for certain silicones often found as residues in finished products.



Substance Name:
Octamethylcyclotetrasiloxane (D4)
EC Number: 209-136-7
CAS Number: 556-67-2

**MEMBER STATE COMMITTEE
 SUPPORT DOCUMENT
 FOR IDENTIFICATION OF
 OCTAMETHYLCYCLOTETRASIOXANE (D4)
 AS A SUBSTANCE OF VERY HIGH CONCERN
 BECAUSE OF ITS PBT¹ AND vPvB² PROPERTIES
 (ARTICLE 57D&E)**

Adopted on 13 June 2018

¹ PBT means persistent, bioaccumulative and toxic
² vPvB means very persistent and very bioaccumulative

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A mathematical model of wetting was developed

Materials and Design 187 (2020) 108389



ELSEVIER

Contents lists available at ScienceDirect

Materials and Design

journal homepage: www.elsevier.com/locate/matdes

How roughness controls the water repellency of woven fabrics

Alain M. Jonas^{a,*}, Ronggang Cai^a, Romain Vermeyen^a, Bernard Nysten^a, Myriam Vanneste^b, David De Smet^b, Karine Glinel^a

^aInstitute of Condensed Matter and Nanosciences, Université catholique de Louvain, Louvain-la-Neuve, Belgium

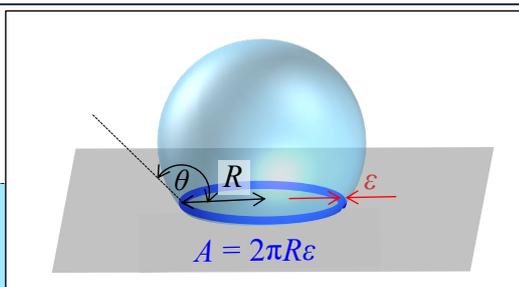
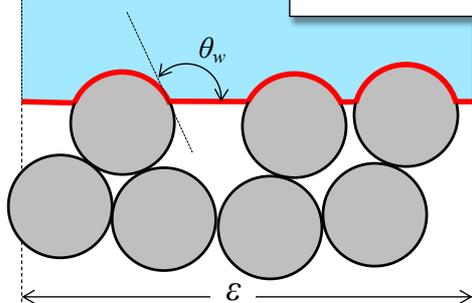
^bCentexbel, Zwijnaarde, Belgium

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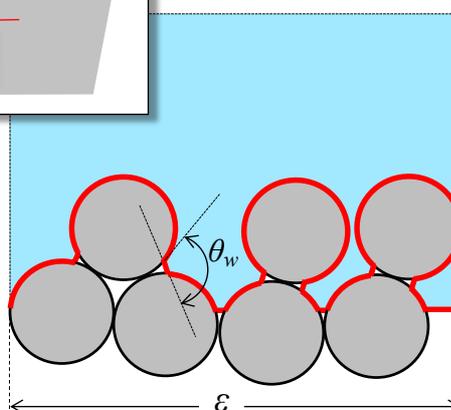
Two main wetting regimes exist

Superficially-wet

water

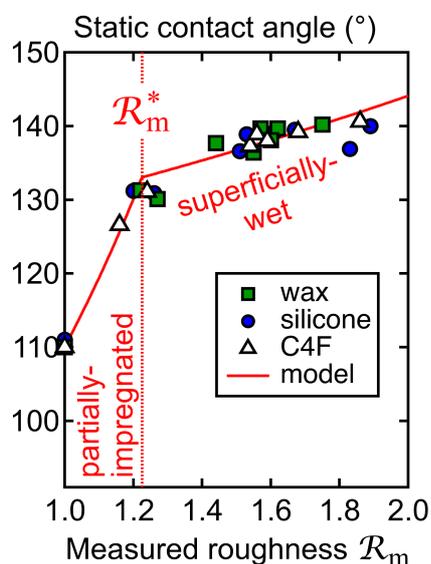


Partially-impregnated



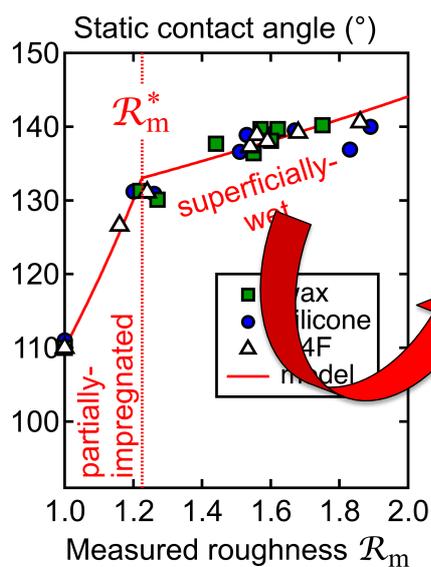
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Contact angles can be predicted from roughness, and give similar results for our three coatings



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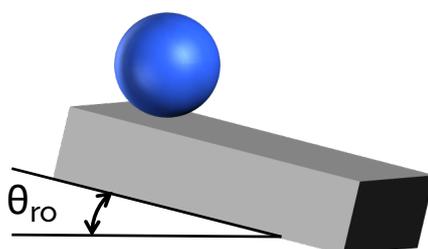
Contact angles can be predicted from roughness, and give similar results for our three coatings



Higher roughness
=> higher amount of trapped air
=> larger contact angle

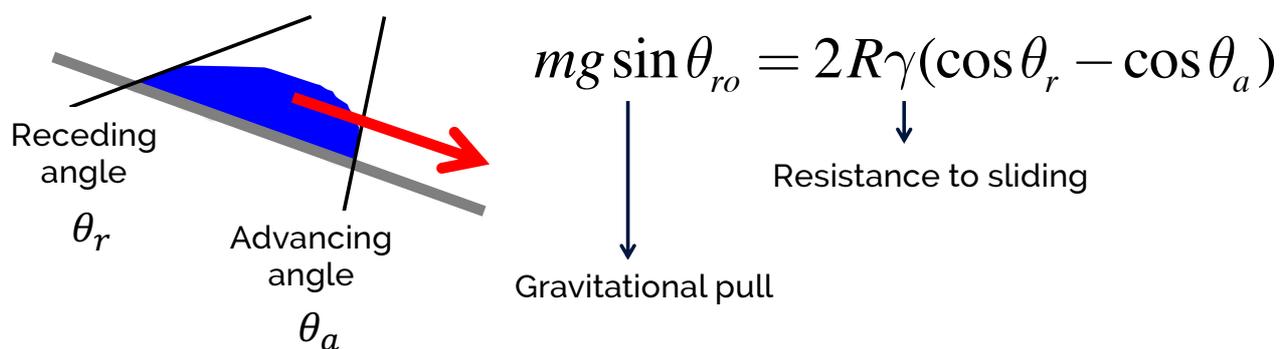
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Roll-off/sliding angle of coated woven fabrics



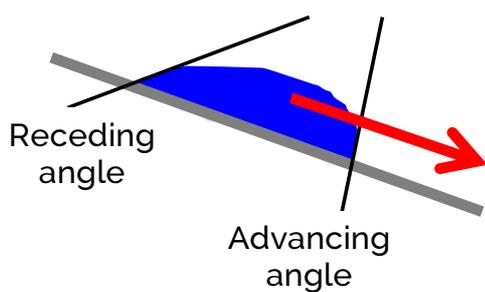
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The roll-off/sliding angle is linked to the pinning of the droplet on the surface



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The roll-off/sliding angle is linked to the pinning of the droplet on the surface



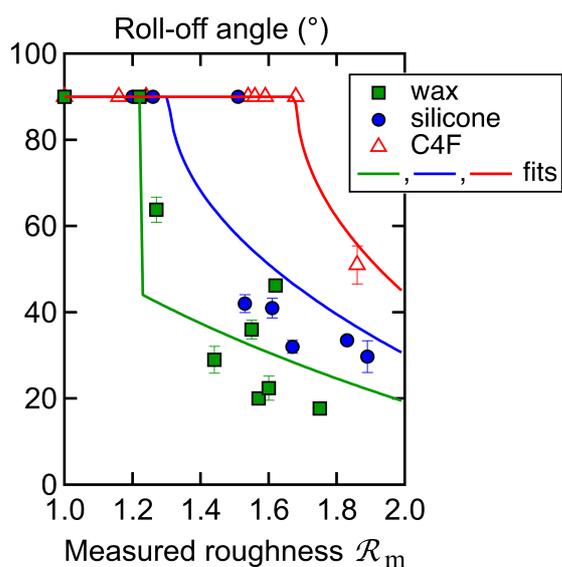
$$mg \sin \theta_{ro} = 2R\gamma(\cos \theta_r - \cos \theta_a)$$

$$= 2R\mu$$

Pinning parameter (N/m)

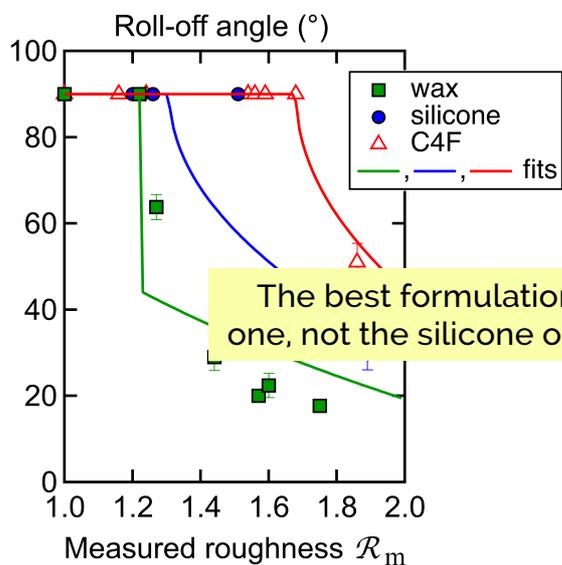
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The roll-off angles discriminate the different coatings



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The roll-off angles correlate to roughness and to the pinning parameter

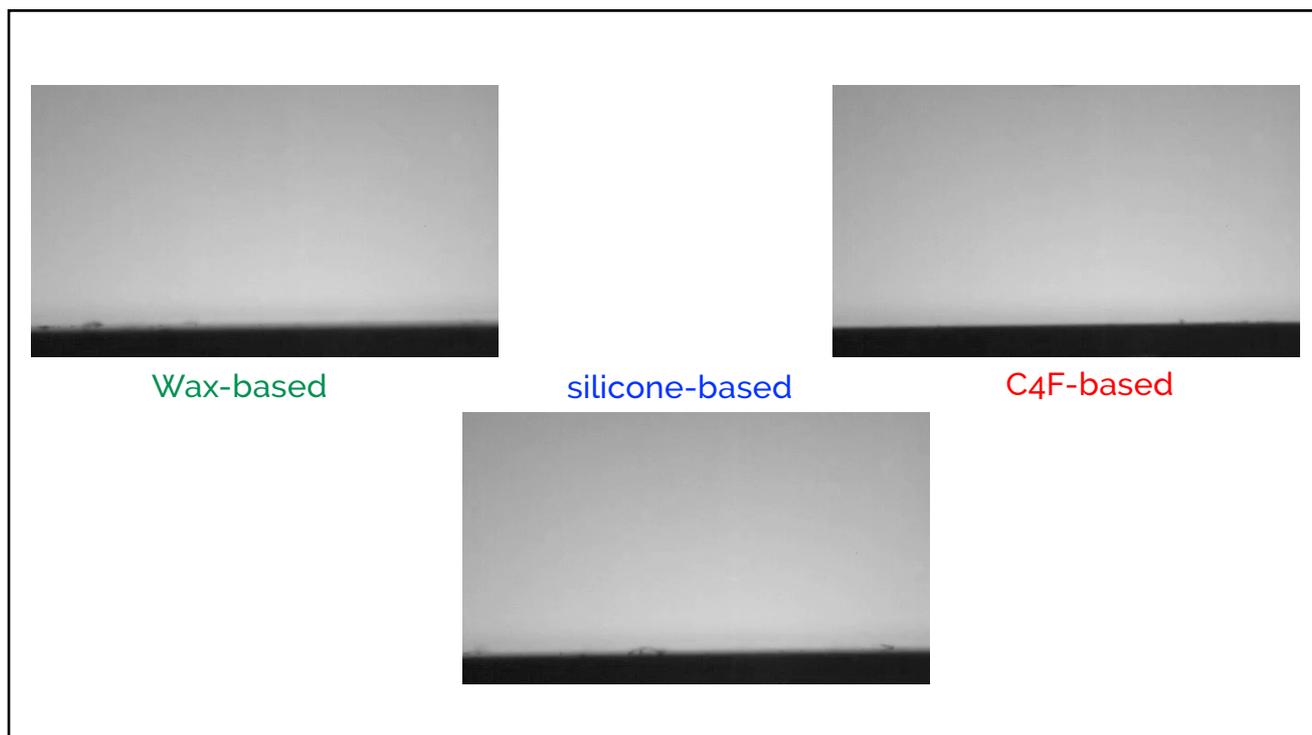


Formulation	μ (N/m) meas.
wax	0.024
silicone	0.029
C4F	0.086

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Future work: bouncing droplets

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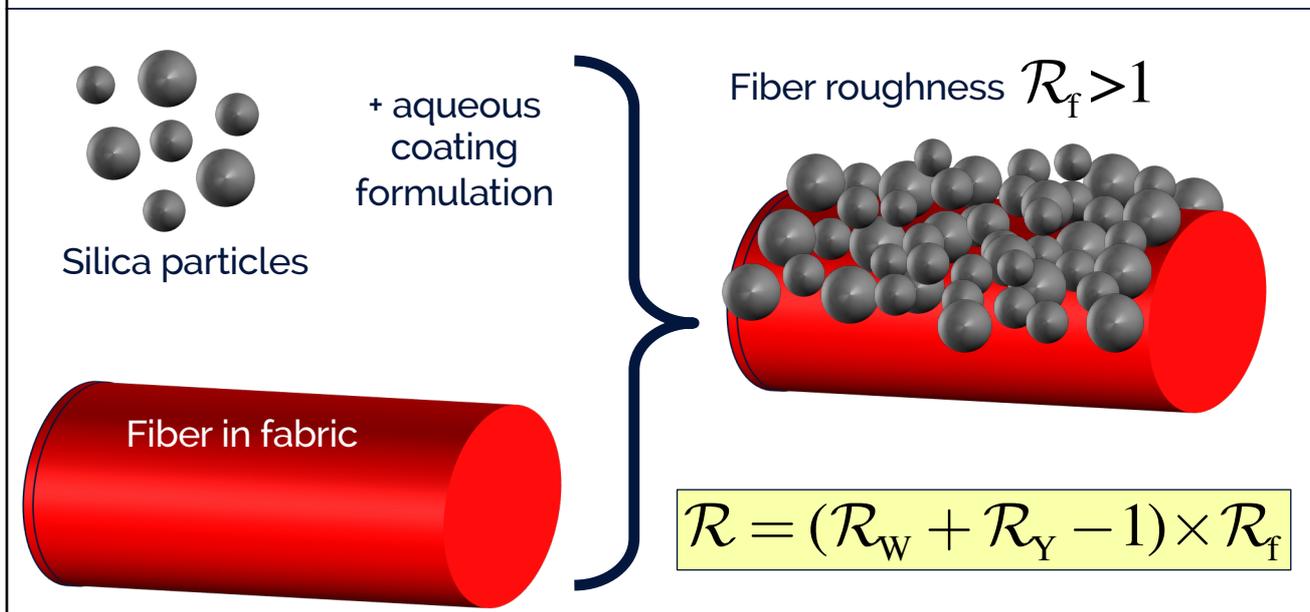


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Can we improve the hydrophobicity
of a fabric of low roughness?

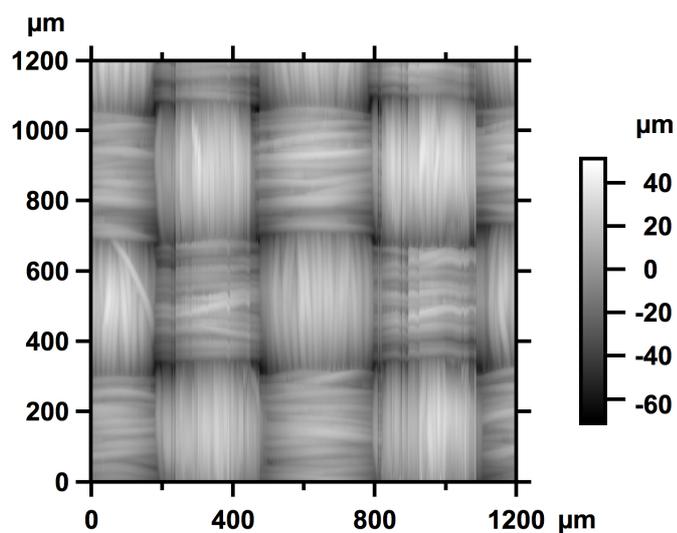
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Boosting fiber roughness with silica particles



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We selected a fabric of low roughness (1.3)



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Results obtained with
a crosslinked silicone rubber
(aqueous emulsion, HC303, Wacker)

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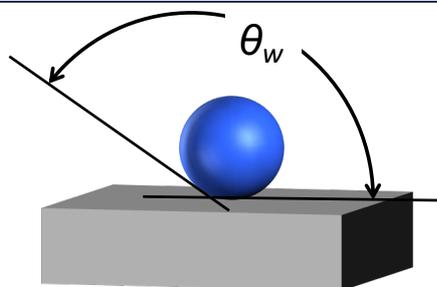
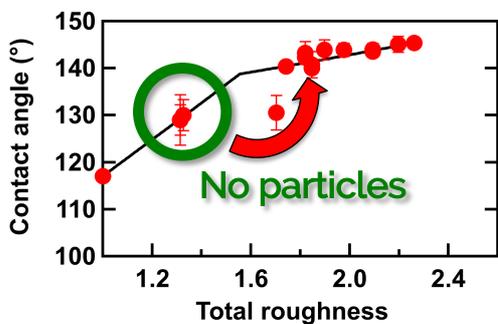
Silicone-based superhydrophobic coatings (dip-coating from aqueous suspensions)



ACS Applied Materials & Interfaces (2018, 10, 15346–15351)

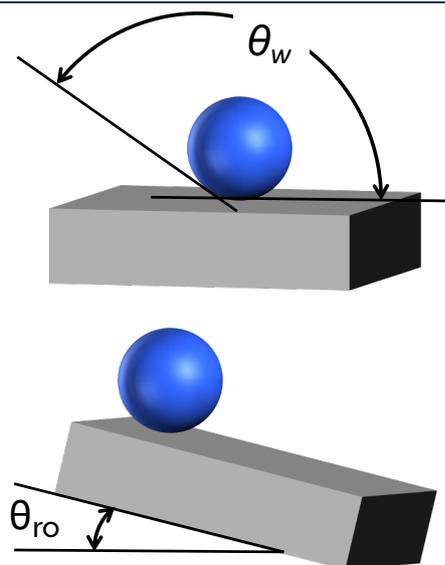
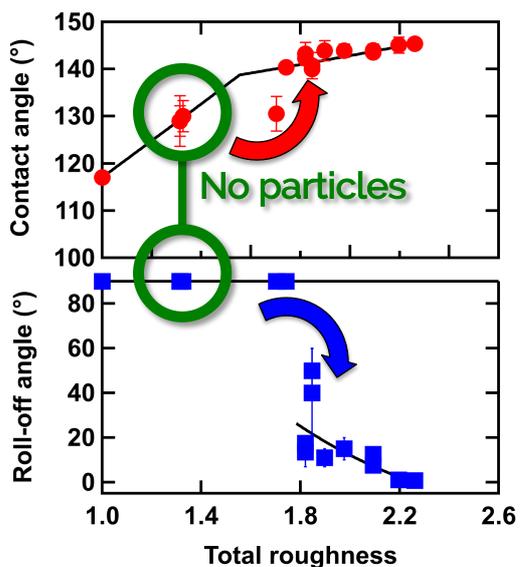
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Increasing the roughness of the fibers results in increased water repellence



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Again, increasing the roughness of the fibers results in increased water repellence



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Water repellence of a silicone-based superhydrophobic fabric prepared from aqueous suspensions



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Future work: oleophobicity

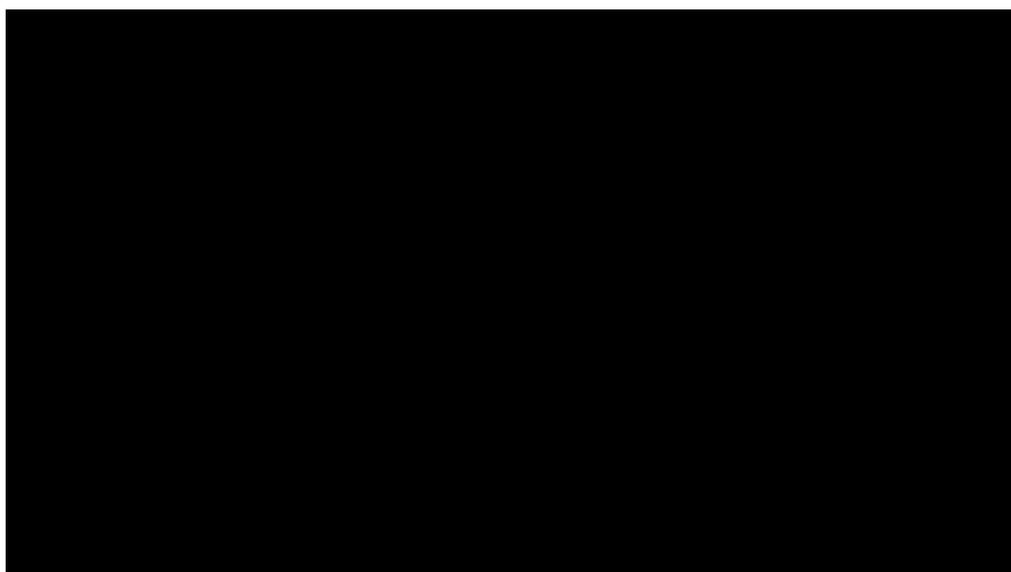
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The short C₄ perfluoroalkyl chains also provide resistance to some organic solvents, with roughened fibers



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The short C₄ perfluoroalkyl chains also lead to superoleophobicity with roughened fibers

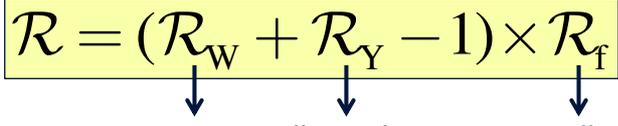


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Main conclusions

1. Total roughness and pinning are predictors of water repellence performance (*other parameters may have to be considered*)

2. Total roughness $\mathcal{R} = (\mathcal{R}_W + \mathcal{R}_Y - 1) \times \mathcal{R}_f$



Weave pattern (limited) Fibers-in-yarn (important) Fiber surface (very important)

3. Fiber surface roughness can be boosted by silica nanoparticles; *other methods certainly exist*

4. Different formulations can be discriminated by the roll-off angle measured on a set of fabrics of different roughness

5. Our studies suggest: waxes > silicones > C4 perfluoroalkyls

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Acknowledgments



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Avec le soutien du Fonds européen
de Développement Régional
Met de steun van het Europees
Fonds voor Regionale Ontwikkeling

Special thanks to:

UCLouvain: Arnaud Delcorte,
Claude Poleunis, Cécile
D'Haese, Romain Vermeyen

Certech: Nicolas Mannu &
Benoît Kartheuser



met de steun van
west-vlaanderen
de gedreven provincie



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